

## Examination and characterization of nanostructured $\text{Co}_{0.9}\text{Ho}_{0.1}\text{MoO}_4$

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**Institute for Technology of Nuclear and Other Raw Mineral Materials**

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decolorization of methylene blue, while their photoelectrochemical activity for water splitting were tested through linear sweep voltammetry in different electrolytes.

## P15

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The nanostructured powder was synthesized by the glycine nitrate procedure (GNP) because it proved to be the most effective and simplest method suitable for controlling the composition and morphology of  $\text{Co}_{0.9}\text{Ho}_{0.1}\text{MoO}_4$ . To prepare nanostructured  $\text{Co}_{0.9}\text{Ho}_{0.1}\text{MoO}_4$ , metal nitrates, and glycine were mixed in appropriate stoichiometric ratios. For the preparation of the technologically important  $\text{Co}_{0.9}\text{Ho}_{0.1}\text{MoO}_4$ , the combustion process has proven to be a promising method that achieves control of stoichiometry, homogeneity, and purity. The synthesized samples were examined by DTA, X-ray diffraction (XRD), Fourier transform infrared (FT-IR) spectra, Spectroscopy, Field emission scanning electron microscopy (FESEM), and nitrogen adsorption method. The acquired nanopowder showed a tendency for agglomeration, inhomogeneous microstructure, and plate-like crystals. The photocatalytic activity of the obtained  $\text{Co}_{0.9}\text{Ho}_{0.1}\text{MoO}_4$  nanopowders was evaluated by the photocatalytic degradation of crystal violet in an aqueous solution. After photocatalytic testing and all the above-mentioned characterizations, it was shown that these nanostructured materials represent promising solutions in photocatalytic processes toward green chemistry and sustainable development.

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